

# **REINHOLD ENVIRONMENTAL Ltd.**



## **2012 NO<sub>x</sub>-Combustion Round Table & Expo Presentation**

February 13-14, 2012, in Columbus, OH / Hosted by AEP

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# **NO Oxidation and Capture with Wet and Dry Scrubbers**

**Bob Crynack – FMC**  
**Sterling Gray - URS**

**2012 RE NOx and Combustion  
Round Table and Expo**

**PerNOxide is a mark of FMC Corporation**

# Agenda

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- Overview of Post Combustion NO<sub>x</sub> controls
- Overview of Oxidation Technologies
- FMC PerNO<sub>x</sub>ide Technology
- Wet Chemistry Research
- Dry Chemistry Research
- Economic Analysis
- PerNO<sub>x</sub>ide Advantages & Challenges
- Applications
- Summary

# Post Combustion Technologies

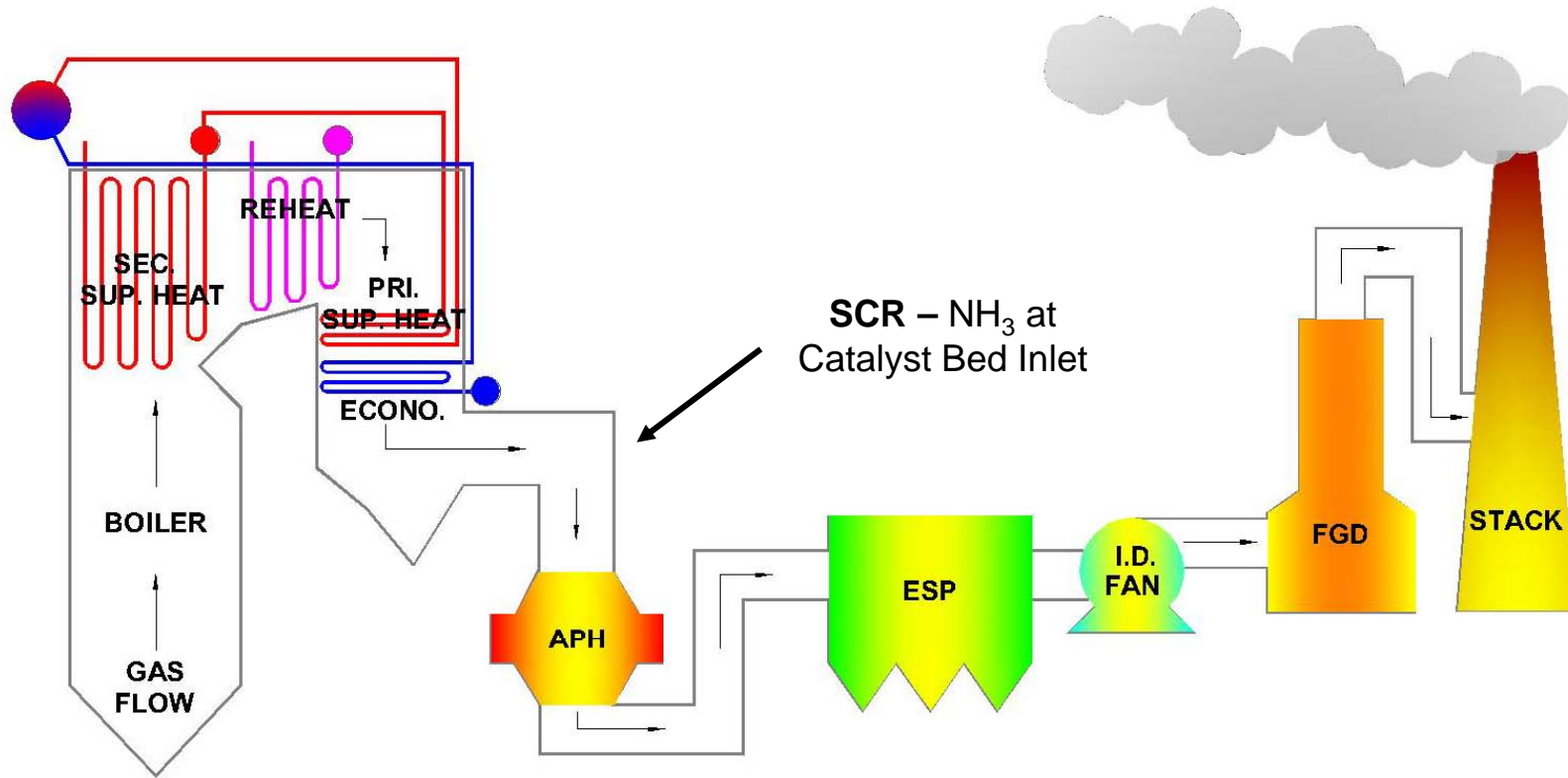
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- Selective Catalytic Reduction - SCR
- Selective Non-Catalytic Reduction – SNCR
- Others, including FMC's PerNOxide

# SCR

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# NO<sub>x</sub> Control Options - SCR



# SCR

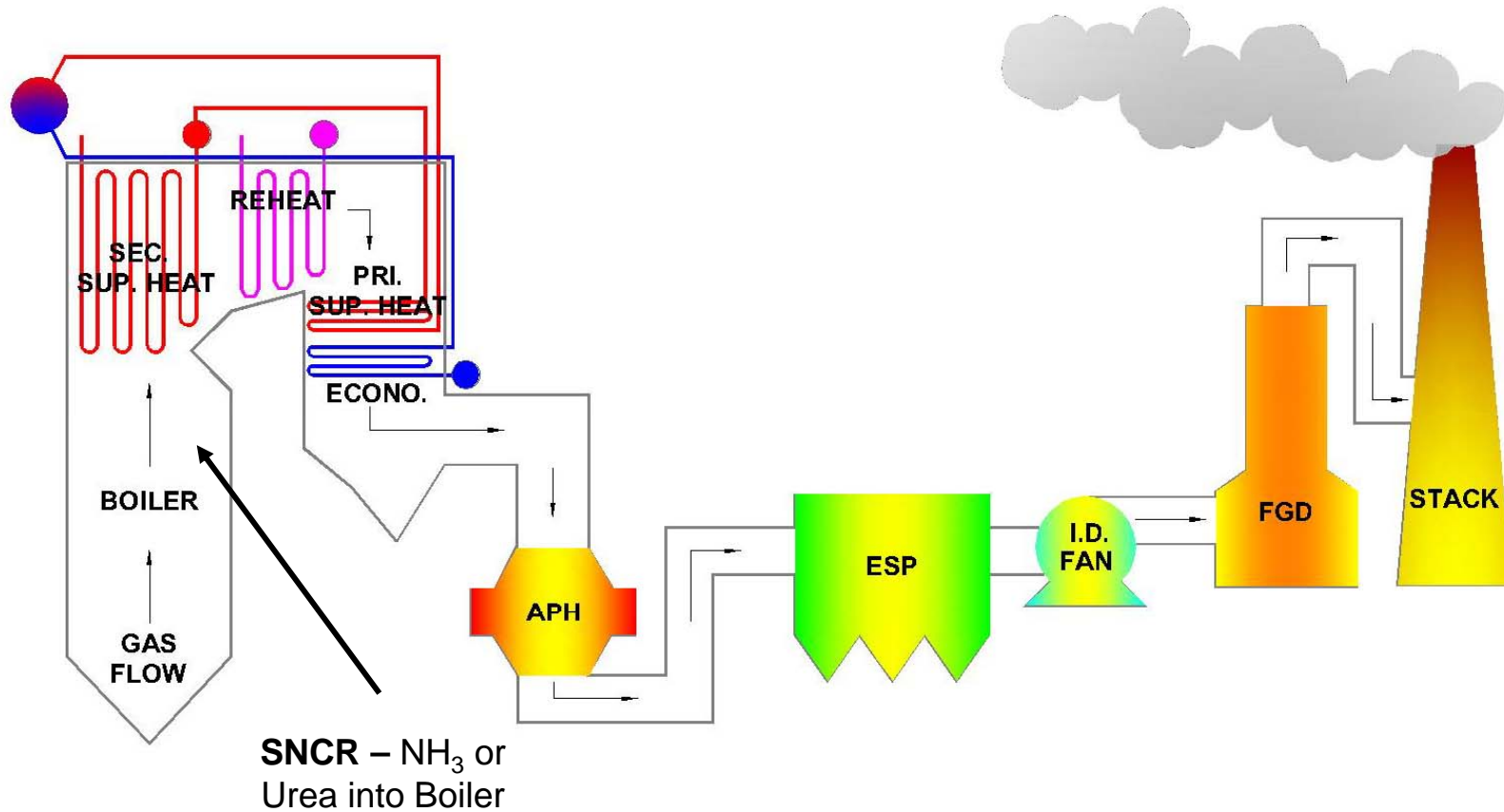
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- Inject  $\text{NH}_3$  ahead of catalyst
  - Anhydrous or aqueous
- NO chemically reduced to  $\text{N}_2$
- $4\text{NO} + 4\text{NH}_3 + \text{O}_2 \rightarrow 4\text{N}_2 + 6\text{H}_2\text{O}$
- $\text{NO} + \text{NO}_2 + 2\text{NH}_3 \rightarrow 2\text{N}_2 + 3\text{H}_2\text{O}$
- NOx reductions of 75-90%
- Excess  $\text{NH}_3$  can react with  $\text{SO}_3$  to form ammonium bisulfate ( $\text{NH}_4\text{HSO}_4$ )
  - Causes air preheater and catalyst fouling
- Catalyst pluggage (ash) and poisoning (As)

# SNCR

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# NO<sub>x</sub> Control Options - SNCR



# SNCR

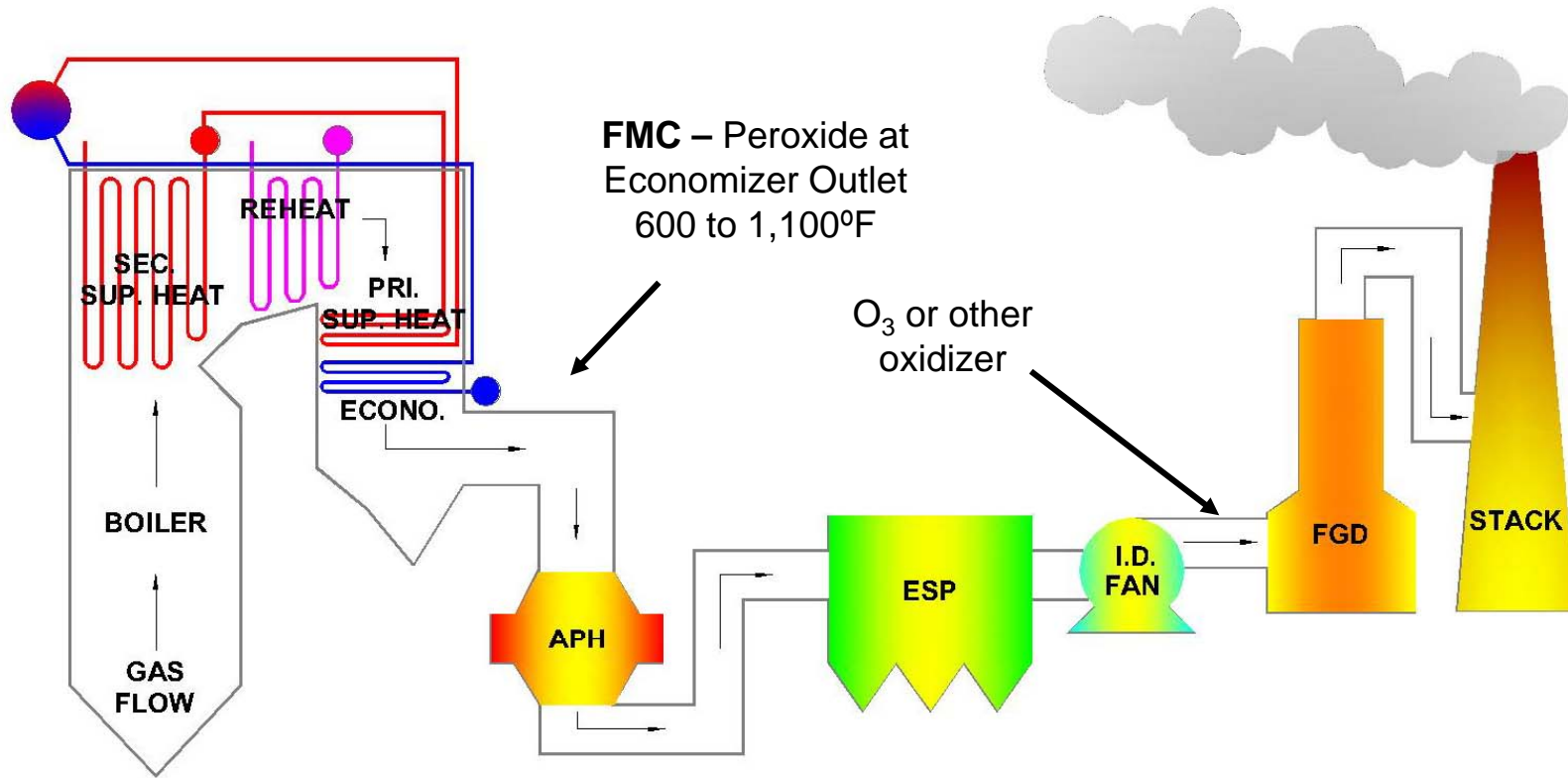
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- Injection at high temperatures in boiler
- Injection temperature is critical - ~1400 - 2000 °F
- Usually multiple injection levels to get close to temperature “sweet spot” under varying operating conditions
- Urea or NH<sub>3</sub> injected. No catalyst needed
- $4\text{NO} + 4\text{NH}_3 + \text{O}_2 \rightarrow 4\text{N}_2 + 6\text{H}_2\text{O}$
- $2\text{NO} + (\text{NH}_2)_2\text{CO} + 1/2\text{O}_2 \rightarrow 2\text{N}_2 + 2\text{H}_2\text{O} + \text{CO}_2$
- NO<sub>x</sub> reductions of 15-40%

# Oxidation Technologies

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# NO<sub>x</sub> Control Options - Oxidation



# NOx Capture by Wet Scrubber

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- Not new. Considered since 1970s when scrubbers became popular and technologists were looking for “multi-pollutant” use of scrubbers
- NO is not water soluble
- Oxidation of NO can lead to many N species
  - $\text{NO}_2$ ,  $\text{N}_2\text{O}_5$ ,  $\text{HNO}_3$ ,  $\text{HNO}_2$
- $\text{NO}_2$  is water soluble, but less than  $\text{SO}_2$
- $\text{N}_2\text{O}_5$  reacts with  $\text{H}_2\text{O}$  to form  $\text{HNO}_3$ , which is highly water soluble

# Ozone (O<sub>3</sub>) Oxidation

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- Ozone generation required
- $\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$
- $2\text{NO}_2 + \text{O}_3 \rightarrow \text{N}_2\text{O}_5 + \text{O}_2$
- $\text{N}_2\text{O}_5 + \text{H}_2\text{O} \rightarrow 2\text{HNO}_3$
- HNO<sub>3</sub> is highly water soluble
- Technologies:
  - LoTOx, Lextran, Comply 2000, Power Span ECO
- All inject at low temperatures
- Claim up to 90% NO<sub>x</sub> reduction

# PerNOxide Process

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- Inject hydrogen peroxide ( $\text{H}_2\text{O}_2$ )
- Normal decomposition:  $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + 1/2 \text{O}_2$
- At elevated temperatures dissociates into
  - $\text{H}_2\text{O}_2 \rightarrow 2 \text{OH} \cdot$
  - $\text{H}_2\text{O}_2 \rightarrow \text{OOH} \cdot + \text{H}$
- Oxidizes NO to  $\text{NO}_2$  and  $\text{N}_2\text{O}_5$
- Can achieve > 80% oxidation
- Up to 70% NOx reduction with wet scrubbers
- Up to 50% NOx reduction with dry scrubbers

# Hydrogen Peroxide (H<sub>2</sub>O<sub>2</sub>)

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- Strong, environmentally friendly oxidizing agent
- Major end uses: pulp & paper, chemicals, food, hair treatment, antiseptic, and electronics
- Product provided in various grades and concentrations
- North American Capacity – 850,000 tons/yr
  - FMC is a leading North American supplier

# Peroxide Characteristics

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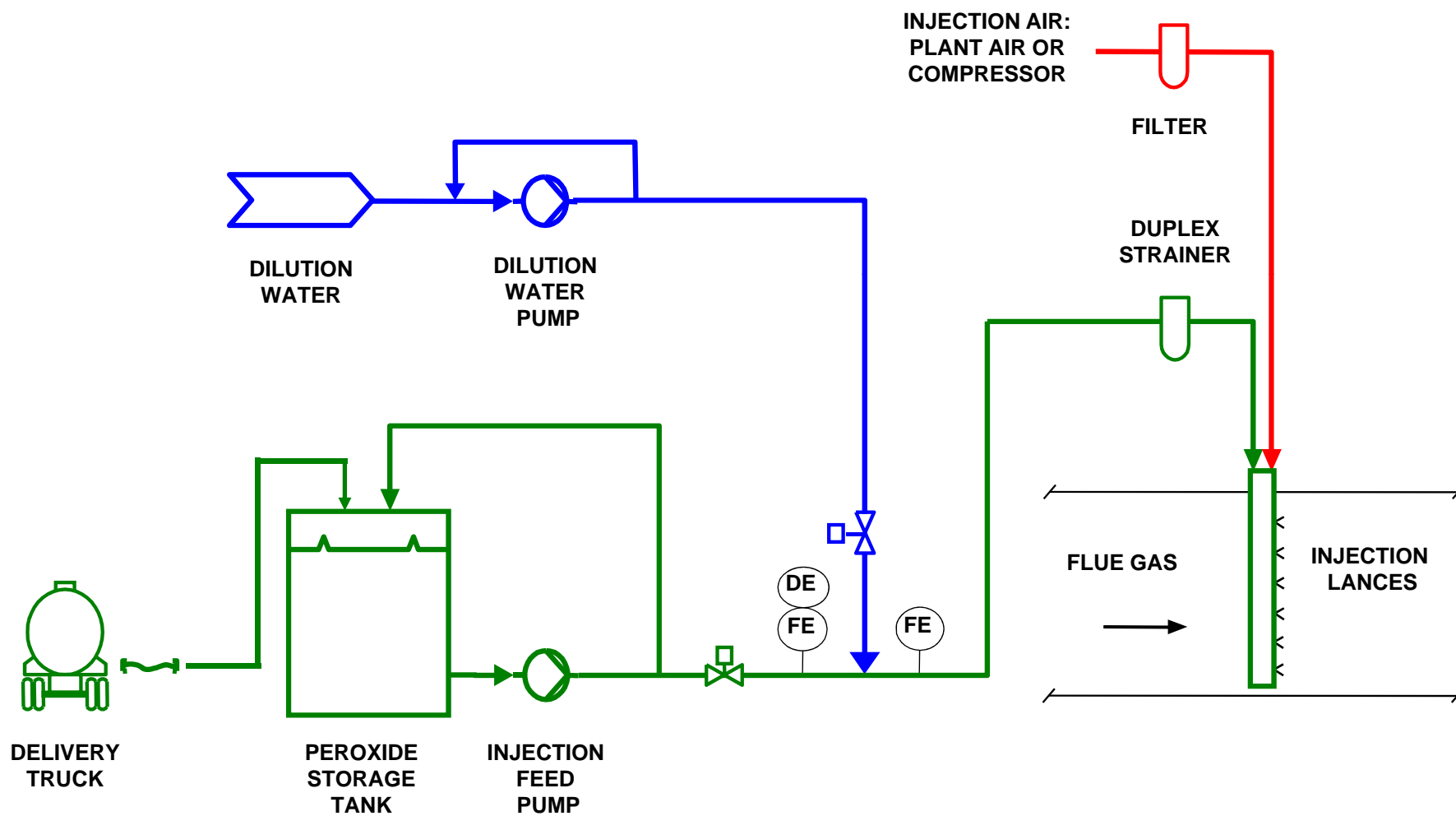
- Specific gravity: 1.2
- Weight: 10 #/gal @ 50% concentration
- Apparent pH:  $\leq 3.0$
- Freezing point:  $-62\text{ }^{\circ}\text{F}$  @ 50% concentration

# Peroxide Injection

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- Injected into duct between economizer and APH inlet
- Concentration controlled to achieve required NO<sub>x</sub> reduction
- Two fluid nozzles (air assisted atomization) used to produce fine droplets
- Water evaporates and peroxide dissociates
- NO oxidized

# Simplified Process Flow Diagram



# Injection Lances

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# Injection Ports

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# NO<sub>x</sub> Capture Options

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- NO<sub>x</sub> Capture Enhancement
  - Wet lime / limestone / sodium scrubbers
  - Circulating Dry Scrubbers (CDS)
  - Spray Dryer Absorbers (SDA)

# PerNOxide Process Development

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- NASA Patent 6,676,912
  - Use of peroxide to oxidize NO to soluble species
  - FMC has exclusive license from NASA
- 3 full scale demonstrations
- FMC & URS Collaboration
  - Process development
  - Technology demonstrations
  - Commercial applications
- FMC testing at EERC on pilot SDA/FF

# Full Scale Demo Conclusions

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- High NO oxidation up to 80%
- Disappointing NO<sub>x</sub> reduction ~20%
- Need to understand wet scrubber chemistry

# URS Wet Chemistry Study

# URS Test Program Objectives

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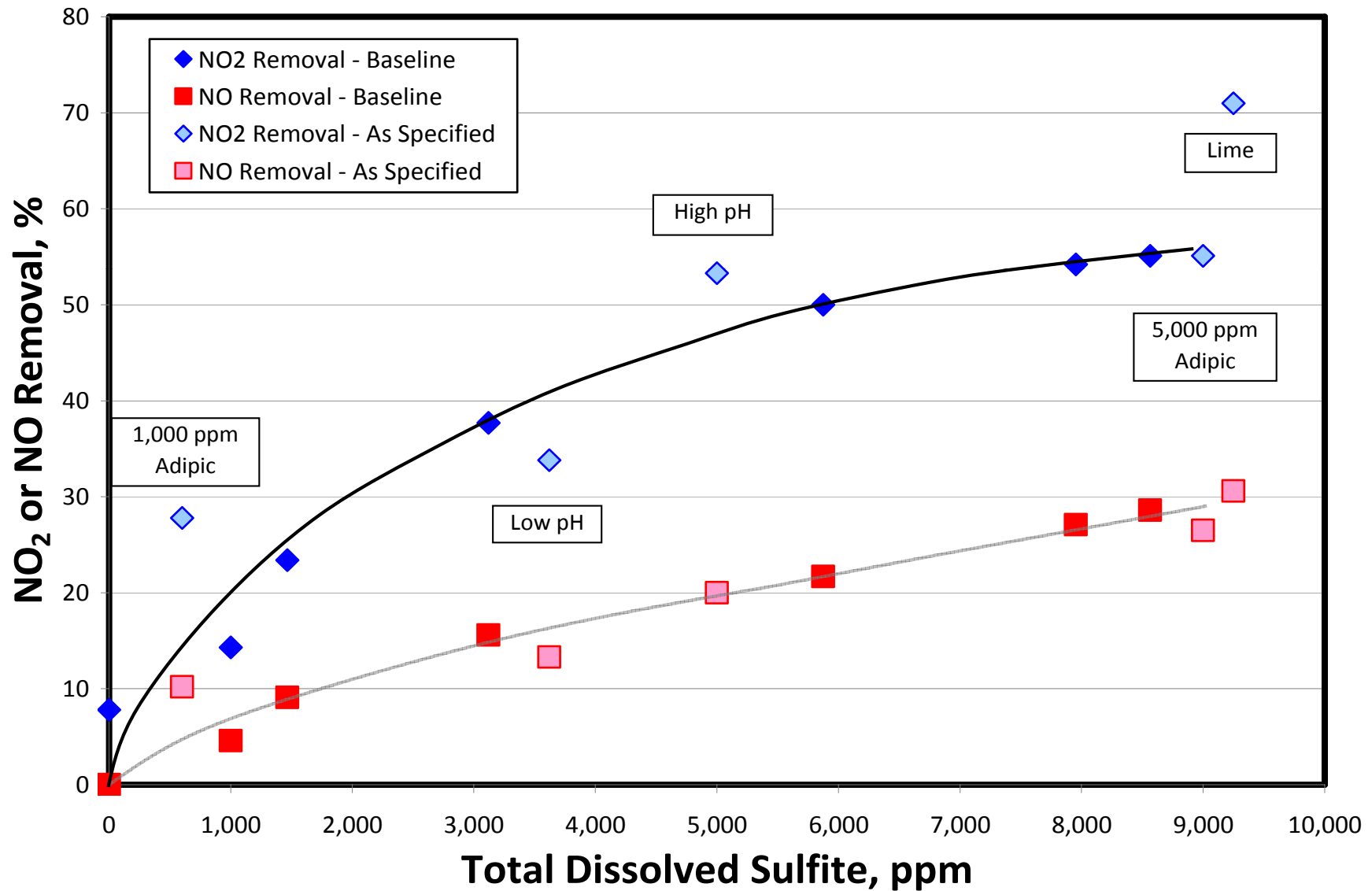
- Conduct Bench-Scale FGD Testing
  - Quantify effect of solution comp. on NO<sub>x</sub> removal
  - Key Variables: Liquid sulfite, pH, buffer additives, thiosulfate
- Perform Computer Modeling of FGD Chemistry
  - Process material balances
  - Aqueous Chemical and Physical Properties Model
  - Investigate how to achieve solution composition necessary for significant NO<sub>x</sub> removal
  - Consider accumulation of dissolved nitrogen species

# Bench-Scale Test Conditions

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- Flue Gas “Bubbler” Contactor
- Contactor Adjusted to Representative Levels
  - Max removal of ~98% SO<sub>2</sub>; ~ 70% NO<sub>2</sub>
- Flue Gas Composition:
  - Simulated flue gas (O<sub>2</sub>, SO<sub>2</sub>)
  - 40 ppm NO, 70 ppm NO<sub>2</sub>
- Solution Composition:
  - Dissolved sulfite: 0 – 9,000 mg/l
  - Solution pH: 5 – 7 (11)
  - Adipic acid: 0 – 5,000 mg/l
  - Thiosulfate: 0 – 2,000 mg/l

# NO<sub>x</sub> Removal Efficiencies



# NO<sub>x</sub> Removal in Wet Scrubbers

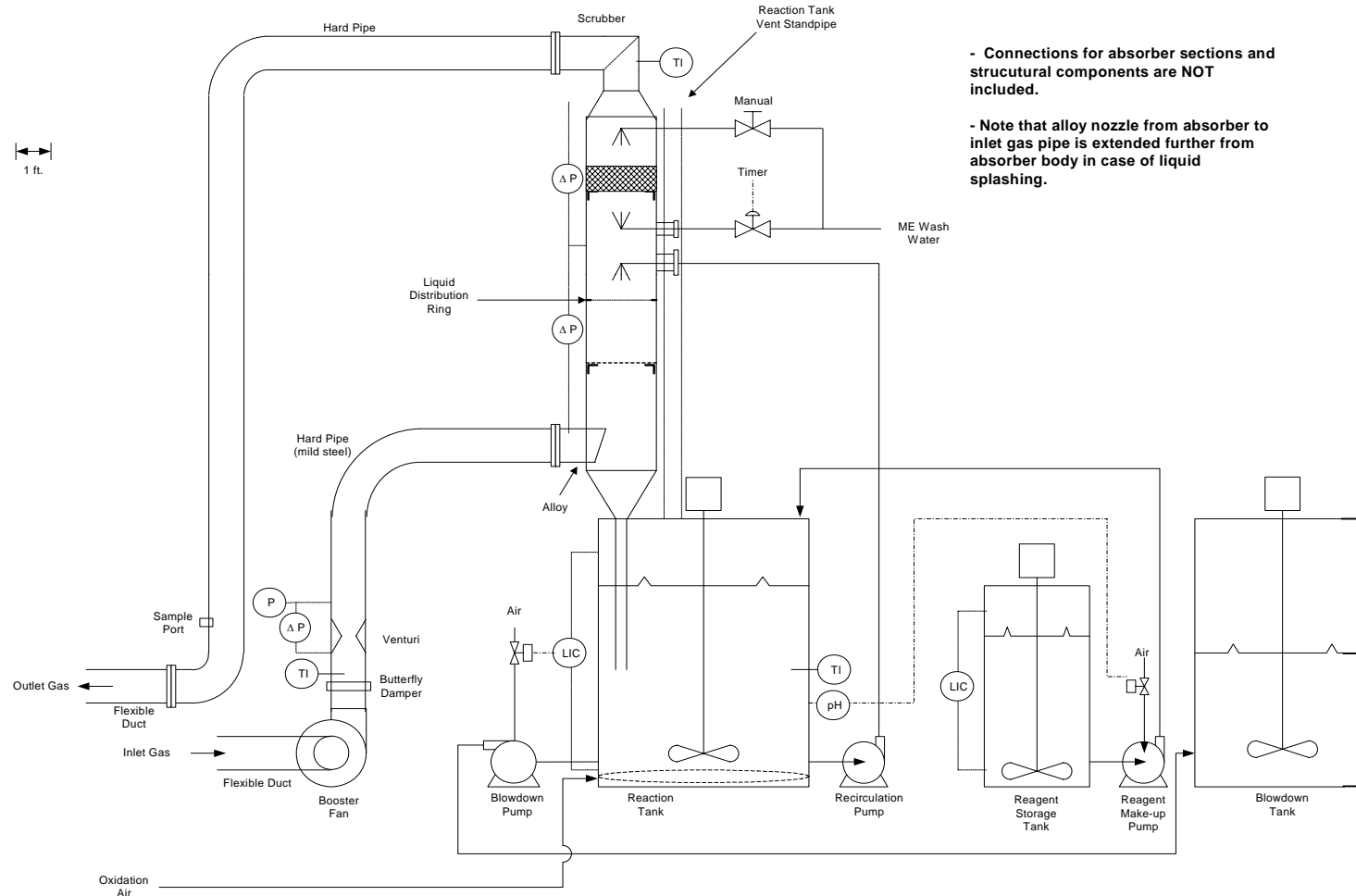
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- Higher-order nitrogen oxides are very soluble with removal efficiencies > 95%
- NO<sub>2</sub> is less soluble, but removal is enhanced by dissolved sulfite
- Reaction products include:
  - Nitrate, S-N species, N<sub>2</sub> gas
- Dissolved sulfite levels must be elevated using scrubber additives:
  - Sodium, magnesium
- Confirmation of bench-scale data needed

# Pilot Wet Scrubber Test Program

- Being conducted in 1<sup>st</sup>Q - 2012
- Use URS pilot wet scrubber (slip-stream)
- NO<sub>2</sub> Addition to simulate Oxidation
- Test NO<sub>2</sub> capture relative to FGD chemistry
  - Sodium
  - Lime, inhibited oxidation
  - Limestone, Inhibited oxidation
  - Limestone, forced oxidation
- Assess longer term impact of N species capture on scrubber chemistry

# Pilot Wet Scrubber Diagram



- Connections for absorber sections and structural components are NOT included.

- Note that alloy nozzle from absorber to inlet gas pipe is extended further from absorber body in case of liquid splashing.

# Pilot Wet Scrubber System

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**FMC**

**URS**

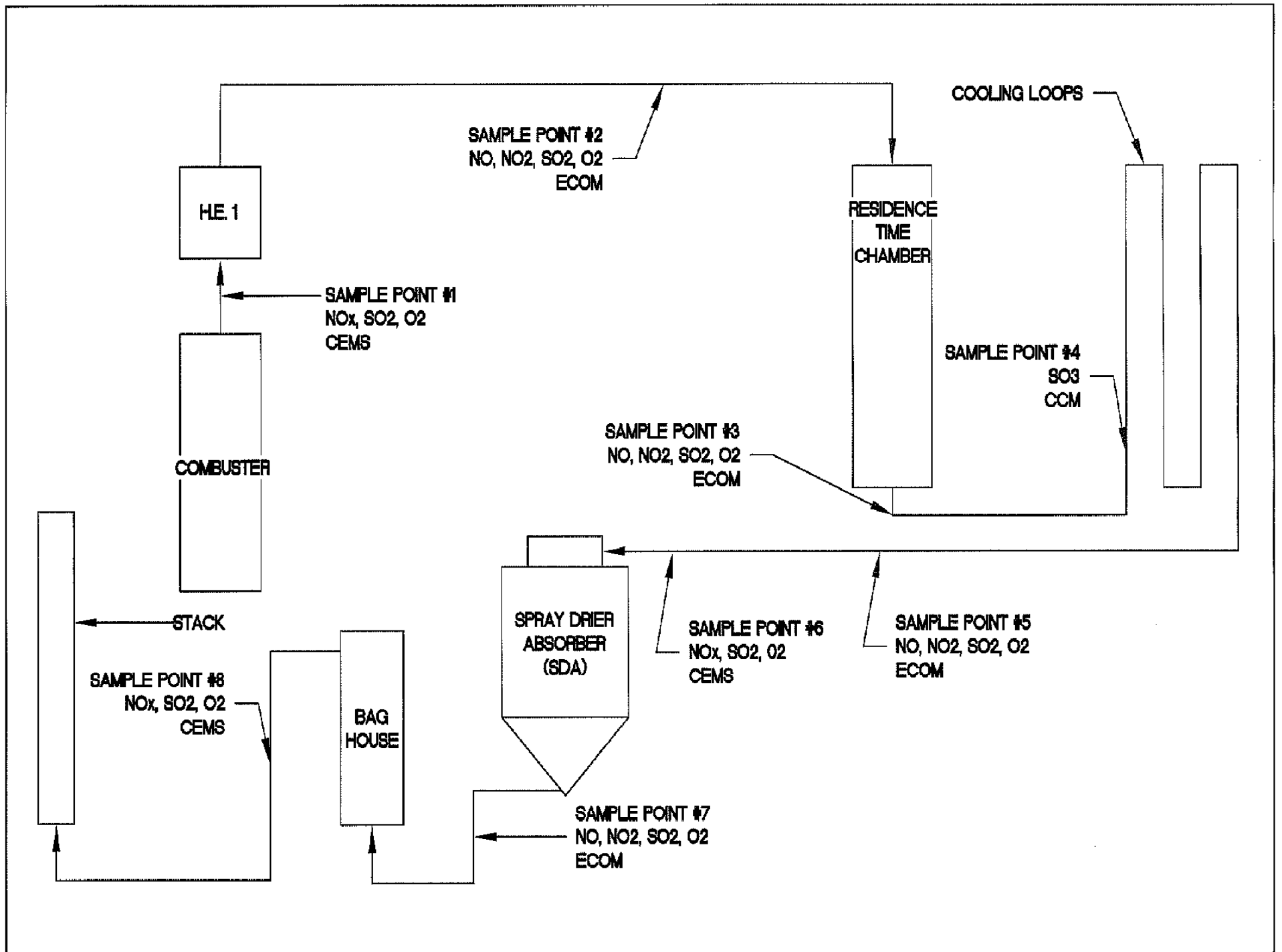
# Dry Chemistry Study

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# Pilot Combustor Specifications

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- Combustor
  - Coal or natural gas
  - 555,000 #/hr, 130 scfm
- Reaction tower
  - Simulate duct residence time
- SDA – rotary atomizer
  - 8 second treatment time
- FF – Pulse jet
  - A/C ratio of 4
  - Teflon bags with Teflon membrane

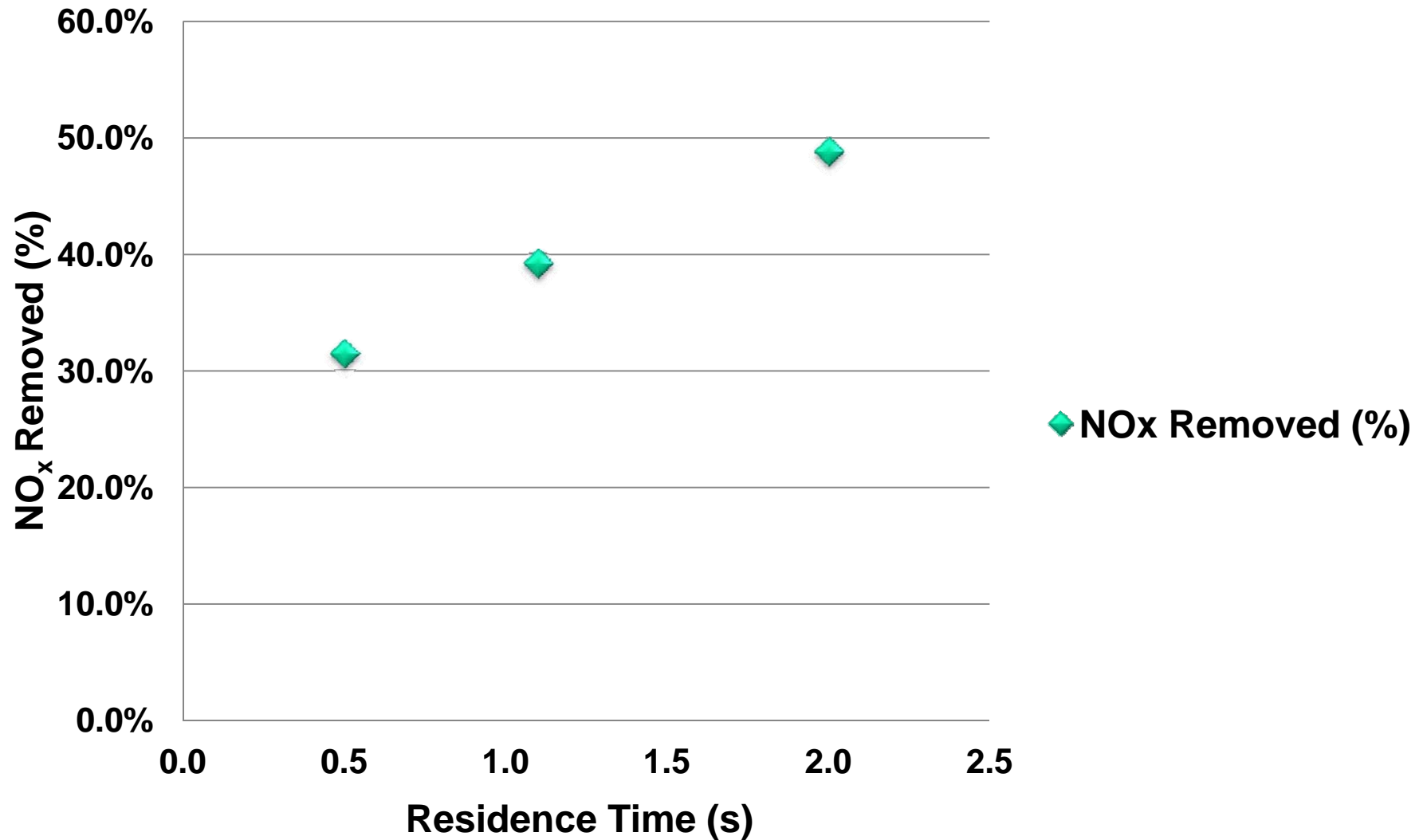


# Purpose of Test Program

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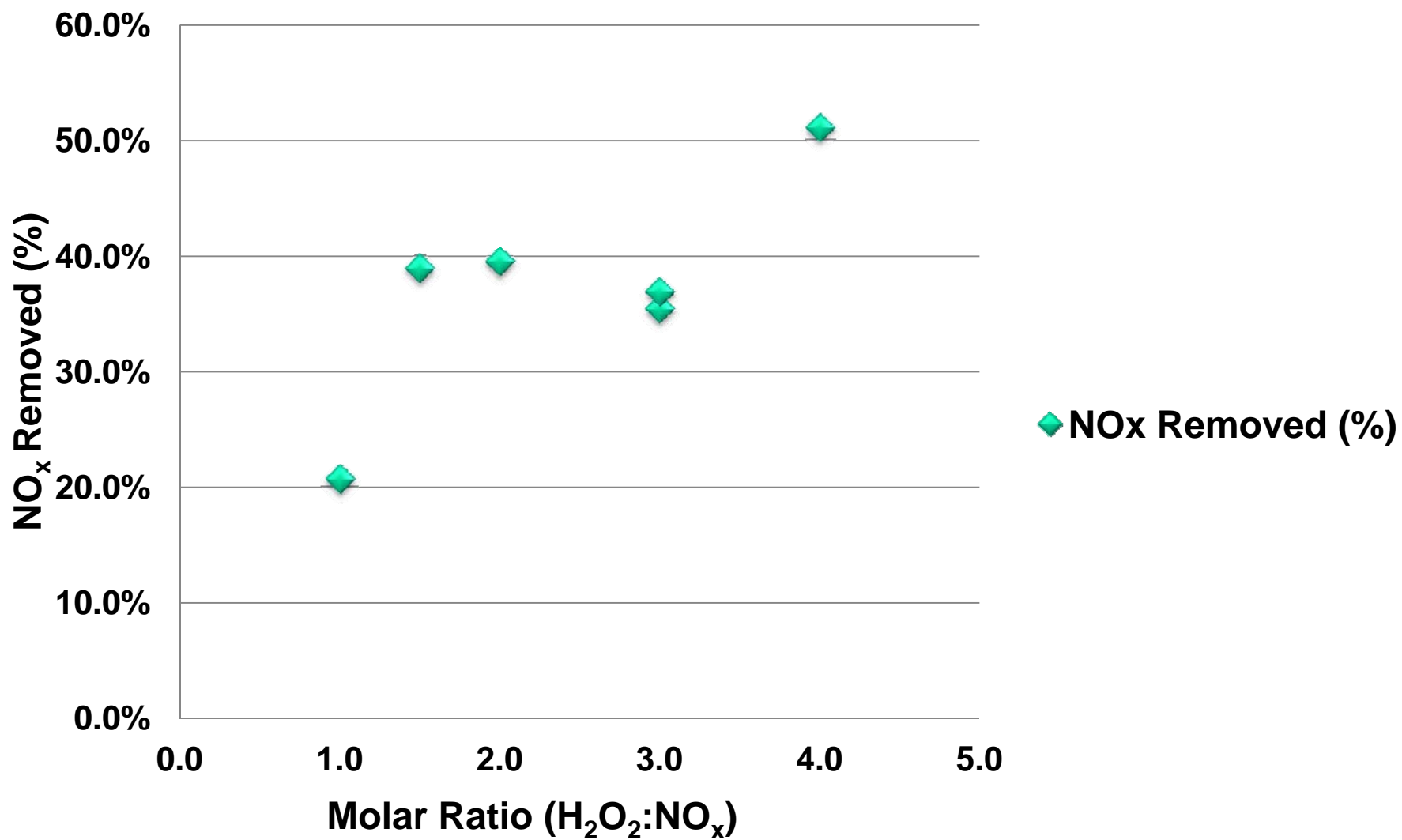
- NOx oxidation
  - Comparison to full scale test programs
- NOx capture
- Parametric study
  - Fuel – Natural gas, Bit, Sub-bit, Lignite coal
  - Treatment time: 0.5 – 2.0 sec
  - Temperature: 600 – 900 °F
  - Peroxide molar ratio: 1.0 - 4.0
  - SO2 concentration: 0 – 2000 ppm

# Residence Time Affect w/ Lignite Fuel @ Baghouse Out (T=800°F, SO<sub>2</sub>=?ppm, MR=1.5)



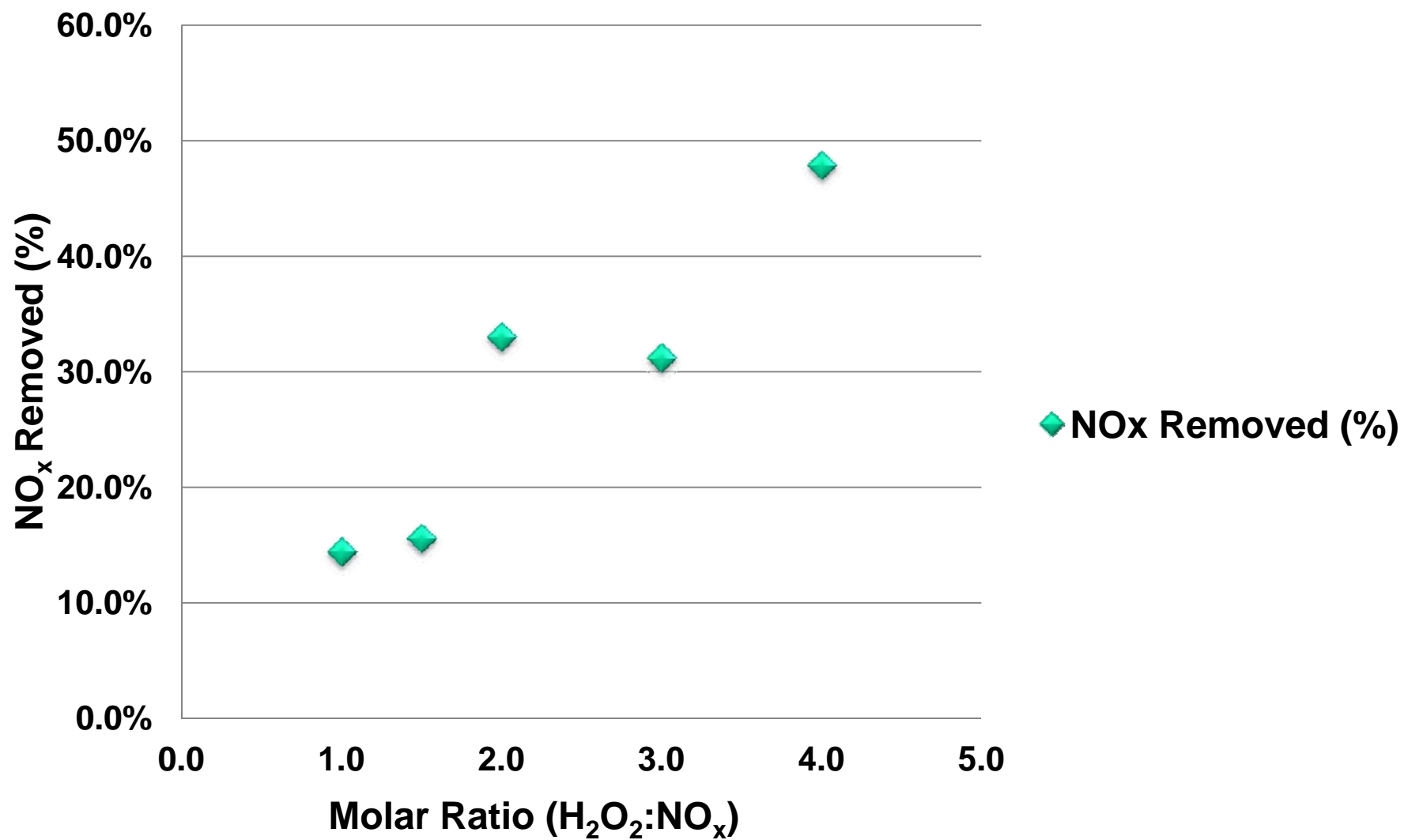
# Molar Ratio Effect w/ E. Bit. Fuel @ Baghouse Out

(T=831°F, SO<sub>2</sub>=634ppm, t=2.0s)



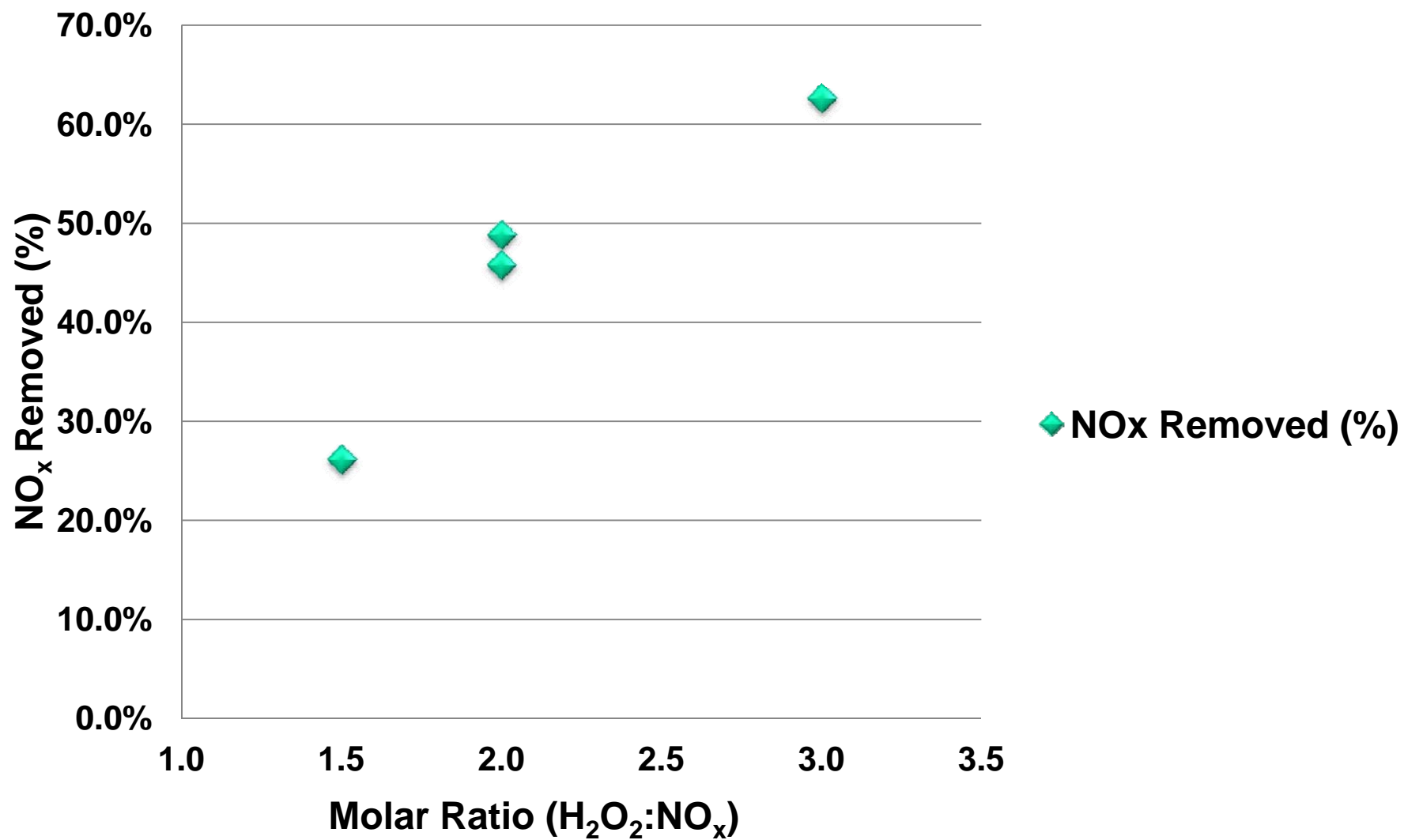
# Molar Ratio Effect w/ PRB Fuel @ Baghouse Out

(T=802°F, SO<sub>2</sub>=731ppm, t=2.0s)



# Molar Ratio Effect w/ Lignite Fuel @ Baghouse Out

(T=800°F, SO<sub>2</sub>=?ppm, t=2.0s)



# Conclusions from EERC Program

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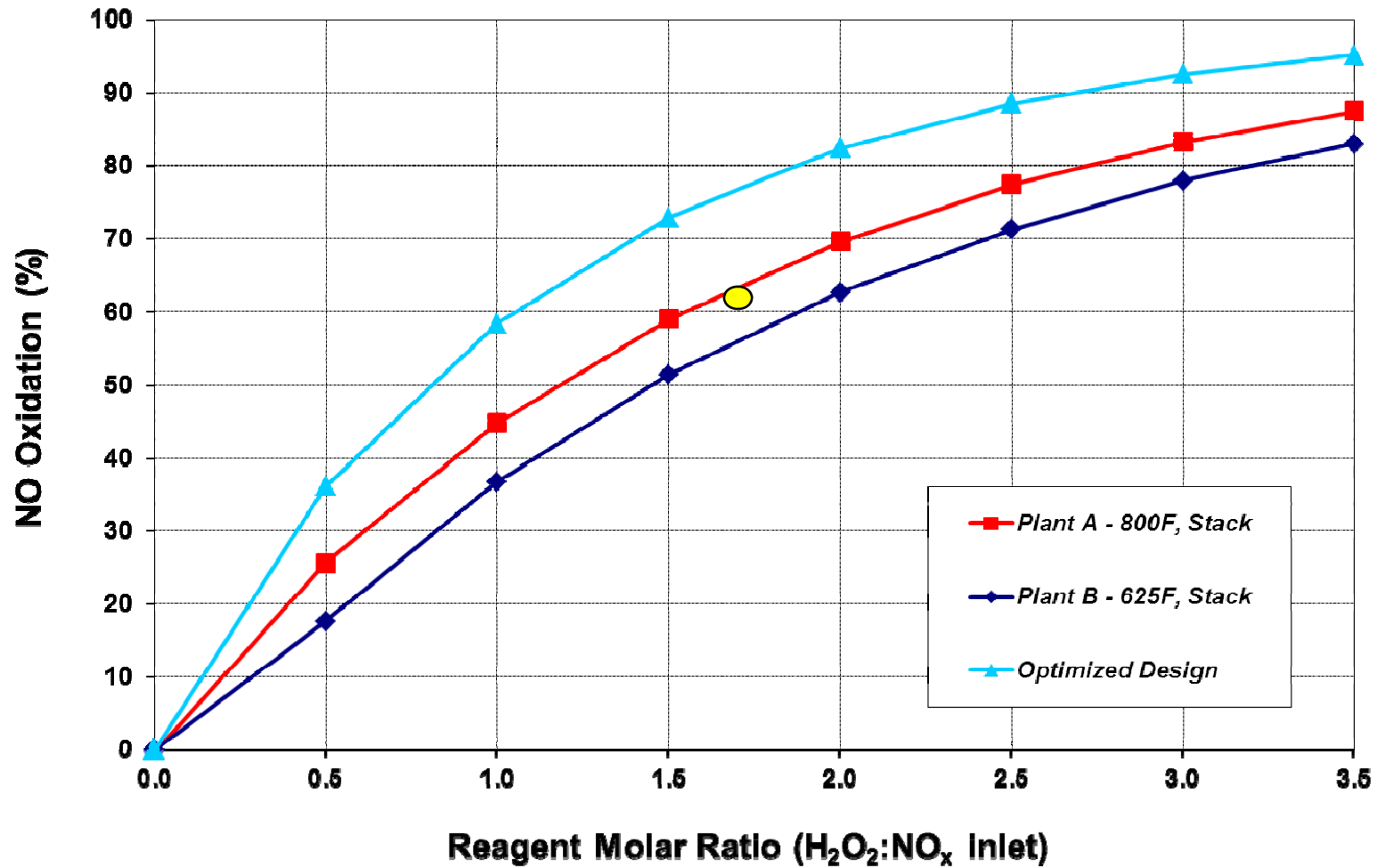
- EERC pilot is a good system to evaluate NOx oxidation and capture across SDA/FF system – limitations of pilot scale
- Better oxidation and capture with
  - Higher temperature
  - Higher SR
  - Longer treatment time
- Variations with fuels – Lignite better
- No effect of SO<sub>2</sub>
- Still need further analysis of this data

# Economic Comparisons

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# NO Oxidation Projections

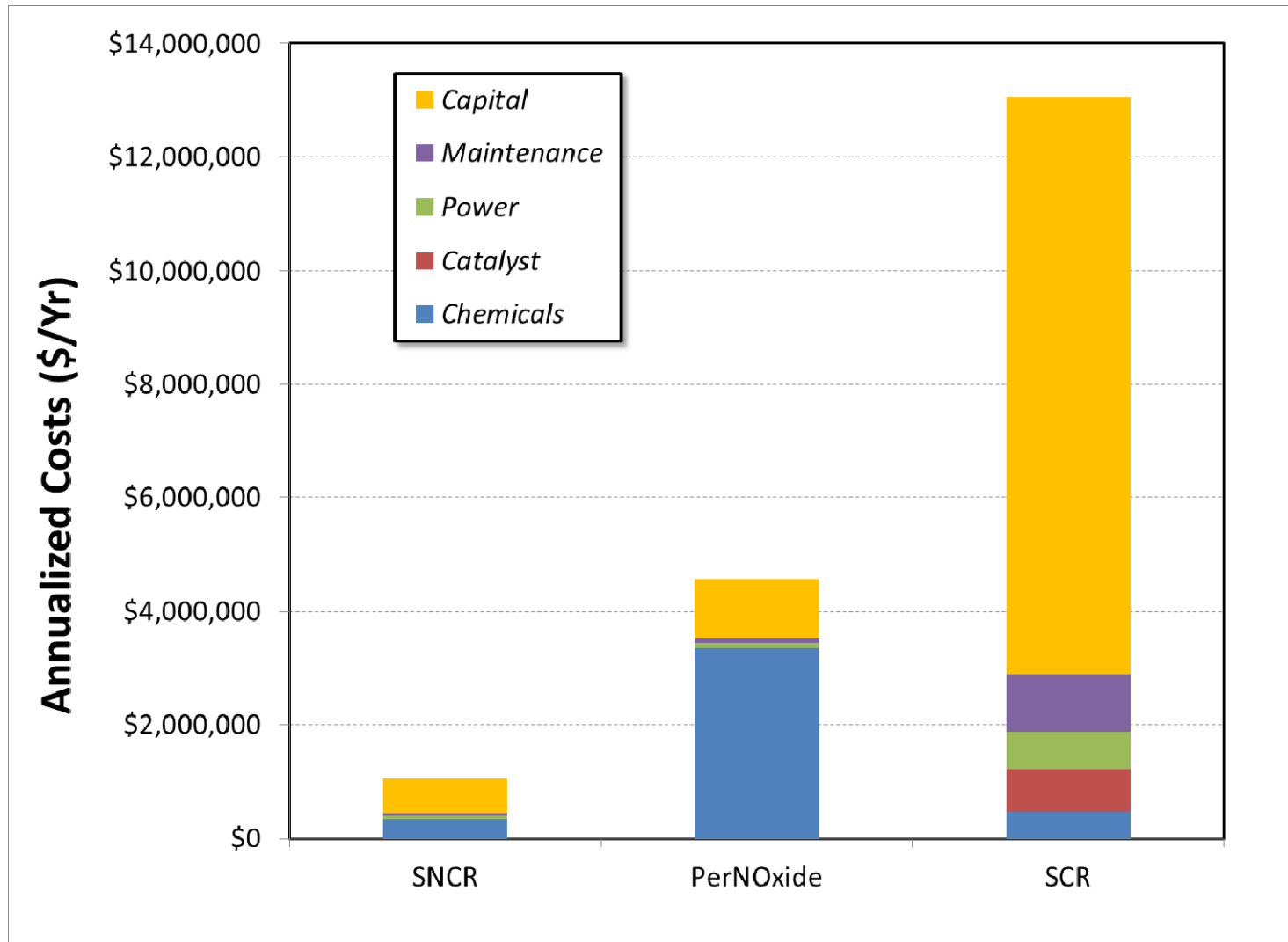
## NO Oxidation vs Reagent Molar Ratio



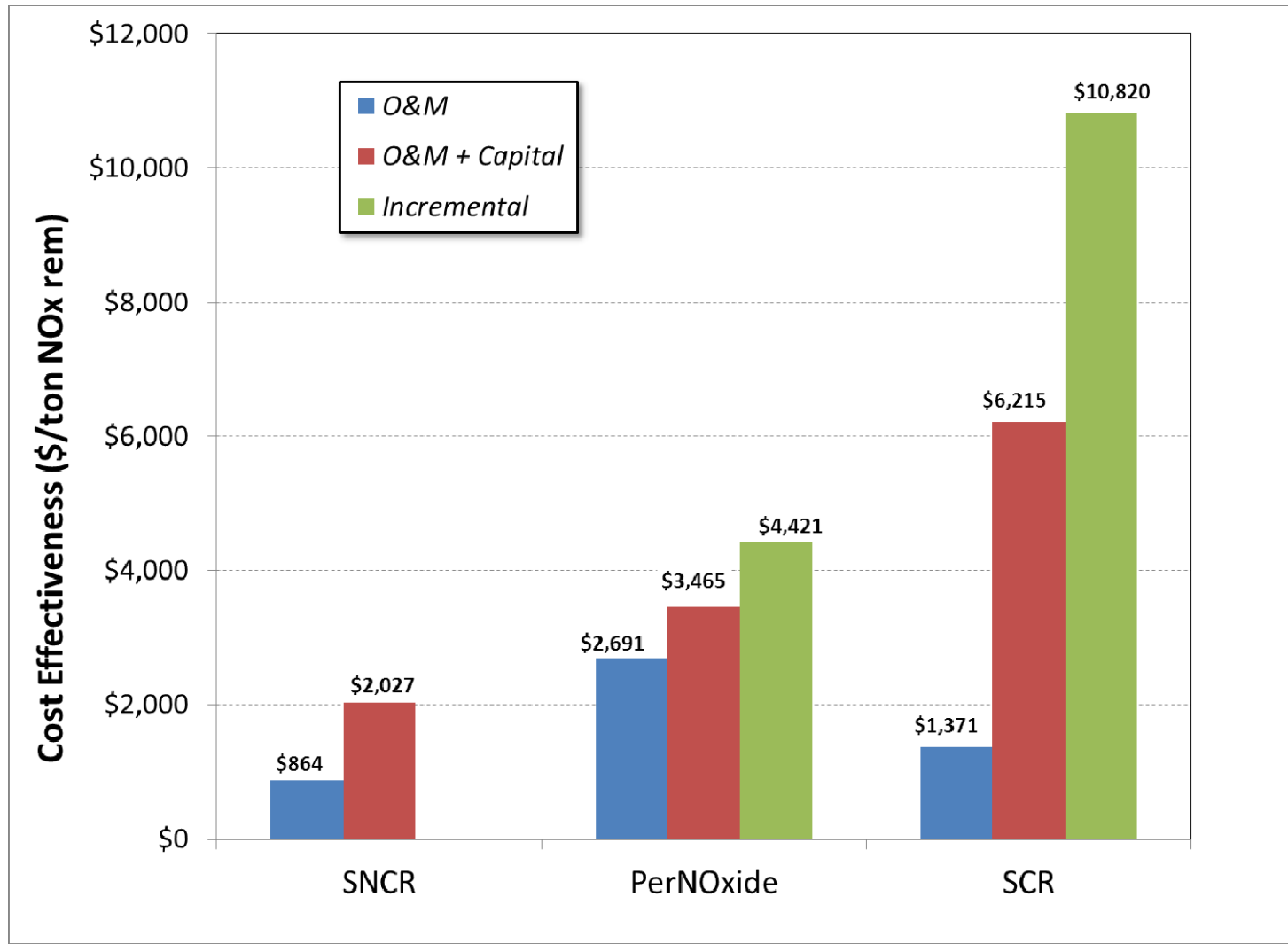
# Technology Cost Comparison

Basis	Units	SNCR	PerNOxide	SCR
Capacity Treated	MW	400	400	400
Inlet NOx	lb/MMBtu	0.20	0.20	0.20
NOx Removal	%	20	50	80
NOx Emissions	lb/MMBtu	0.16	0.10	0.04
NOx Removed	TPY	526	1,316	2,102
Capacity Factor	%	75	75	75
Reagent		Urea	Peroxide	Ammonia
Reagent Molar Ratio	mol:mol NOx	0.20	1.50	0.80
Reagent Cost	\$/ton	\$500	\$1,000	\$600
Soda Ash Cost	\$/ton	\$0	\$300	
Catalyst Cost	\$/c.f.			\$150
SCR Catalyst Life	Yrs			3
Power Cost	\$/MW-hr	\$30	\$30	\$30
Annual Maint. Cost	% of Capital	1.0	1.0	1.0
Capital Cost	\$/kW	\$15	\$25	\$250
Capital Recovery Period	Yrs	20	20	20
Capital Discount Rate	%	8	8	8
Capital Recovery Factor	%	10	10	10

# Annualized Costs



# Cost Effectiveness Analysis



# Take Aways...

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- PerNOxide is cost-effective alternative to SCR for modest NOx reductions
- PerNOxide economics more favorable for ...
  - Lower inlet NOx levels
  - Smaller unit size
  - Lower capacity factor
  - Sodium and lime scrubbers
  - Space constrained site (difficult retrofit)
- PerNOxide can be final “layer” of controls before SCR required...
  - LNB, OFA, SNCR, ....

# PerNOxide Advantages and Challenges

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# PerNOxide Technology Advantages

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- Lower capital & maintenance costs than SCR
- Higher removal efficiencies than SNCR
- Operational flexibility
- Mercury oxidation and capture
- Immune to SCR catalyst poisons
- Environmentally friendly reagent
- Limited downtime for installation
- Project execution < 1 year

# PerNOxide Technology Challenges

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- Developing Technology
- May Require Modification to FGD Chemistry
- Possible Increase in NO<sub>2</sub> Emissions
- Possible Increase in SO<sub>3</sub> Formation
- Waste Stream Management

# PerNOxide Bridges NO<sub>x</sub> Control Gap

	<b>SNCR</b>	<b>SCR</b>
<b>NO<sub>x</sub> Removal</b>	20-30%	75-90%
<b>Capital Costs</b>	Moderate	High
<b>Boiler Impacts</b>	Yes	No
<b>Operational Issues</b>	Ammonia Slip	Ammonia Slip & SO <sub>3</sub> Generation

# PerNOxide Bridges NO<sub>x</sub> Control Gap

	<b>SNCR</b>	<b>PerNOxide</b>	<b>SCR</b>
<b>NO<sub>x</sub> Removal</b>	20-30%	30-70%	75-90%
<b>Capital Costs</b>	Moderate	Low	High
<b>Boiler Impacts</b>	Yes	No	No
<b>Operational Issues</b>	Ammonia Slip	Potential Visible Emissions	Ammonia Slip & SO <sub>3</sub> Generation

# Summary

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- PerNOxide technology is a 2-Step Process
  - Oxidation of NO to NO<sub>2</sub> and other N-O forms
  - Capture of NO<sub>2</sub> and other N-O forms
- High NO oxidation achieved using peroxide
- Capture of NO<sub>2</sub> is critical
  - Bench scale testing to explore NO<sub>2</sub> Removal
  - Key variables: Scrubber mass transfer, dissolved sulfite, pH, & additives
  - Modification of scrubbing liquor composition may be required

# Summary

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- Economics are favorable compared to SCR
- Example 400 MW plant:
  - Capital costs 1/10<sup>th</sup> that of SCR
  - Annualized costs 1/3<sup>rd</sup> that of SCR
  - Cost effectiveness (\$/ton) half that of SCR
  - Incremental SCR costs > \$10,000/ton NO<sub>x</sub>
- Favorable economics driven by:
  - Low capital costs
  - Moderate (30 to 70%) NO<sub>x</sub> removal effectiveness
  - Operational flexibility (NO<sub>x</sub> removal “on demand”)

# Path Forward

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- **FMC and URS seeking opportunities for full scale demonstrations on both wet and dry scrubber applications**

**Thank You!**

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**Questions???**

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